

# Luminescent Sulfophosphate Glass Doped with Plasmonic Nanoparticles: Optical and Spectral Features

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The lanthanide ( $\text{Ln}^{3+}$ ) ions doped sulfophosphate glass systems are promising candidates to be used in different fields, including fiber-optics, waveguide lasers, optical detectors, solar concentrators, and solid-state lasers. The variable and weak interactions between metaphosphate ( $\text{PO}_3^-$ ) and sulfate ( $\text{SO}_4^{2-}$ ) ions are a great feature for rare earth ion incorporation and give these glasses a high luminescence efficiency with the least non-radiative losses. By adding different modifier oxides, such as alkaline earth oxides, the interaction between  $\text{PO}_3^-$  and  $\text{SO}_4^{2-}$  can be greatly improved. It could result in a disturbance in the glass system and consequently facilitate the creation of non-bridging oxygen (NBO) groups [1].

At an optimum  $\text{Sm}^{3+}/\text{Er}^{3+}$  dopant concentration, emission increases by combining the remarkable properties of rare-earth ions and sulfophosphate glass, such as energy transfer between the erbium and samarium ions. For further improvement of the emission properties, incorporation of plasmonic nanoparticles such as  $\text{TiO}_2$  and Ag into the RE-doped glass system has been reported as an effective way to overcome the low emission cross-section of the systems. Spectroscopic techniques represent a unique combination to investigate the structural and optical features concerning the incorporation of  $\text{Sm}^{3+}$  and  $\text{Er}^{3+}$  ions in sulfophosphate glasses, as well as the role of plasmonic nanoparticle incorporation to modify the optical emission properties of such systems [2–5].

In this contribution, the spectroscopic as well as optical properties of  $\text{Sm}^{3+}/\text{Er}^{3+}$  doped glass, including titania and silver nanoparticles, are presented. Erbium ( $\text{Er}^{3+}$ ) and samarium ( $\text{Sm}^{3+}$ ) ions doped magnesium zinc sulfophosphate glass systems containing  $\text{TiO}_2$  NPs were synthesized using the melt-quenching technique and characterized for the influence of varying  $\text{TiO}_2$  ion contents on their spectral properties. UV-Vis-NIR spectra revealed the occurrence of several absorption bands corresponding to the transitions from the ground state to various excited states of the ions. An enhancement in the PL emission intensities in the presence of  $\text{TiO}_2$  or Ag NPs were observed. Such enhancement was attributed to the local field effect stimulated by the surface plasmon resonance (SPR) of plasmonic NPs and the energy transfer mechanism among  $\text{Ln}^{3+}$  ions and the NPs.

## References

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